

Discovery of suboxidic coordinate in high- $T_{\rm C}$ ferromagnetic semiconductor Co-doped TiO₂

High-Curie-temperature diluted ferromagnetic semiconductors (high- $T_{\rm C}$ DFSs) are strongly desired for the room-temperature operation of semiconductor spintronic devices. Among them, high- $T_{\rm C}$ DFS oxides have been intensively studied since the discovery of room-temperature ferromagnetism in Co-doped TiO₂ in 2001 [1]. The $T_{\rm C}$ of Co-doped TiO₂, ~600 K, is extraordinarily high considering the conventional carrier-mediated ferromagnetism seen in GaMnAs (~200 K). However, the microscopic origin of the high- $T_{\rm C}$ ferromagnetism has remained a mystery for a long time.

X-ray fluorescence holography (XFH) is a powerful tool for resolving a local structure, enabling a 3D atomic environment to be visualized around a selected element within a radius of nanometer orders without assuming any prior models [2]. To date, the advantageousness of XFH has been demonstrated for the evaluation of local lattice distortions around specified elements [3] and for the detection of nanocluster-like structures [4]. In this work, we experimentally determined local structures around the Co atoms in paramagnetic and ferromagnetic rutile Co-doped TiO₂ films by measuring XFH holograms. In addition, first-principles calculations based on density functional theory were carried out to evaluate the validity of the structure model obtained by XFH. In the ferromagnetic Co-doped TiO₂ film, suboxidic coordination around Co embedded in the rutile structure was observed. This suboxidic coordination may generate a ferromagnetic embryo leading to the strong exchange interaction realizing the high- $T_{\rm C}$ in Co-doped TiO₂ [5].

Two rutile $Ti_{1-x}Co_xO_2$ films with x = 0.01 and 0.05 were used, where the former and latter are paramagnetic and ferromagnetic, respectively. The XFH experiment was carried out at beamline BL39XU, and XAFS measurements were performed at beamline BL22XU. XFH data were recorded at incident X-ray energies of 8.0-12.0 keV in 0.5 keV steps by measuring Co K α fluorescence (6.9 keV) intensity variation as a function of the incident beam direction. Figure 1(a) shows the hologram pattern of the Ti_{0.95}Co_{0.05}O₂ taken at 8.0 keV. The hologram data were symmetrized using the mirror symmetry of rutile TiO₂ (101) and then extended using the rotoreflection symmetry as shown in Fig. 1(b). Such data manipulation is important for providing accurate and high-resolution atomic images. The 3D atomic arrangements around the fluorescent element Co



Fig. 1. Co holograms of $Ti_{0.95}Co_{0.05}O_2$ film. (a) Raw data and (b) symmetrized data. In (b), the data are mirror-symmetrized with respect to the dashed circle and then rotoreflection-symmetrized with respect to the (100) plane.

were reconstructed from multiple-energy holograms using Barton's algorithm [6].

The 3D atomic images of Ti_{0.99}Co_{0.01}O₂ and $Ti_{0.95}Co_{0.05}O_2$ are shown in Figs. 2(a) and 2(b), respectively. In Fig. 2(a), the reconstructed 3D atomic images are superposed on the ideal Ti atomic positions of the rutile TiO₂, as marked by open circles, while the images of O atoms are hardly visible owing to their low scattering powers. Since the rutile structure, which shows in Fig. 2(c), is most stable form of TiO₂, the visible images are contributed to Ti atoms. Therefore, the Co atoms in the paramagnetic Ti_{0.99}Co_{0.01}O₂ substituted for Ti atoms without a large lattice distortion from the rutile structure. In contrast, the 3D atomic images in Fig. 2(b) are markedly different from the rutile structure. The surrounding atomic arrangement exhibits fourfold symmetry with respect to the normal to the rutile (101) plane (z-axis). Taking into account the very low visibility of O atoms, we can assume that the O atoms are located just above and below the Co atom along the z-axis, as denoted by dashed circles (Fig. 2(d)), corresponding to CoO₂Ti₄ coordination as discussed below.

To confirm the validity of our structure models in Figs. 2(c) and 2(d), we measured the Co *K*-edge XAFS spectra and calculated the XAFS spectra using the structure models and FEFF. The XAFS spectrum of the $Ti_{0.99}Co_{0.01}O_2$ agrees with that calculated using the rutile structure in Fig. 2(c). The spectrum of the $Ti_{0.95}Co_{0.05}O_2$ was calculated using the structure model in Fig. 2(d). The CoO_2Ti_4 and CoO_2Co_4 structure models were used for the calculation and their XAFS spectra were compared. The CoO_2Ti_4 structure model more closely reproduces the experimental spectrum of the Ti_{0.95}Co_{0.05}O₂ film.

To evaluate the stability of the suboxide cluster in the host rutile TiO₂, we carried out a series of firstprinciples calculations using the VASP code with the HSE06 hybrid functional for several possible atomic configurations. As the first step, to represent an isolated cluster model, the single $CoO_2Ti_4+(Ti_8)$ cluster depicted in Fig. 2(d) was embedded in the rutile supercell structure while keeping the nearest Ti atoms trivalent. The cluster soon changed into an arrangement similar to that of the rutile crystal during the structure optimization process. This result indicates that an isolated CoO₂Ti₄ cluster is very unstable, also supporting the absence of the suboxide cluster in the low Co-doped film. Next, we aligned two CoO₂Ti₄ clusters in the rutile supercell structure $(Co_2O_{64}Ti_{34})$, as shown in Fig. 3(a), with an initial configuration where the O atoms are located above and below the Co atom orthogonal to the cluster alignment, and we obtained the structure in Fig. 3(b) by structure optimization. This structure preserves the original arrangement surrounding the CoO₂Ti₄ cluster, reproducing the XFH result in Fig. 2(b).

In conclusion, the presence of suboxide coordination in rutile $Ti_{0.95}Co_{0.05}O_2$ was revealed by the XFH and XAFS measurements. Possible atomic configurations of the suboxide network were investigated by first-principles calculation, which



Fig. 2. Reconstructed real-space images around Co in (a) $Ti_{0.09}Co_{0.01}O_2$ and (b) $Ti_{0.95}Co_{0.05}O_2$ films. (c) and (d) Structure models obtained from the images in (a) and (b), respectively.

justified the reason why the clusters appear only at a high Co concentration. Furthermore, the suboxide network might be the source of the strong exchange interaction leading to the high $T_{\rm C}$ in Co-doped TiO₂.



Fig. 3. Possible structures of the suboxide cluster embedded in rutile TiO_2 . The structures were obtained by structure optimization based on a first-principles calculation. (a) Two aligned suboxide clusters embedded in rutile TiO_2 . (b) Fully relaxed clusters.

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References

- [1] Y. Matsumoto *et al.*: Science **291** (2001) 854.
- [2] K. Hayashi *et al.*: J. Phys.: Condens. Matter **24** (2012) 093201.
- [3] W. Hu et al.: Phys. Rev. B 89 (2014) 140103(R).
- [4] W. Hu et al.: Phys. Rev. B 80 (2009) 060202(R).
- [5] W. Hu, K. Hayashi, T. Fukumura, K. Akagi, M. Tsukada, N. Happo, S. Hosokawa, K. Ohwada, M. Takahasi, M.
- Suzuki and M. Kawasaki: App. Phys. Lett. **106** (2015) 222403.
- [6] J.J. Barton: Phys. Rev. Lett. 67 (1991) 3106.